STRUCTURE OF ALNUSERIC ACID, THE FIRST REPORTED NATURALLY OCCURRING ${\tt C_{31}}\text{-SECODAMMARANE-TYPE TRITERPENE FROM } \textit{ALNUS SERRULATOIDES CALL. }$

Toshifumi HIRATA, Reiko IDEO, and Takayuki SUGA*

Department of Chemistry, Faculty of Science, Hiroshima University

Higashisenda-machi, Hiroshima 730

A novel C_{31}^{-3} ,4-secodammarane-type triterpene, alnuseric acid, was isolated from the male flowers of Alnus serrulatoides Call. (Betulaceae). Its structure was elucidated to be (20s, 24R) - 20, 24-epoxy-24-methyl-3,4-secodammar-4(28)-en-3-oic acid (I) by a combination of chemical and spectroscopic methods.

In connection with the chemical and physiological studies of pollen, $^{1)}$ we have reported the triterpenoids and phytosterols of the male flowers of Alnus serrulatoides Call. (Japanese name: Kawara-hannoki). $^{2\sim4)}$ We now have isolated a new 31 -secodammarane-type triterpene, named alnuseric acid (I), from the male flowers. Naturally occurring triterpenes having a seco-form have been well documented, $^{5)}$ but as yet no 6 -seco-triterpene. We here report evidences which led to the establishment of its structure.

Column chromatography of the ether soluble fraction of the male flowers gave alnuseric acid (I) as colorless needles, mp 107~109°C, [α] $_D^{25}$ +24.3° (c 0.21, MeOH),

I: R=COOH
IV: R=CN

II: R=H III: R=OH

^{*} To whom all inquiries should be addressed.

m/e 472 (M⁺); C₃₁H₅₂O₃ (requires: C 78.76; H 11.09. found: C 79.06; H 11.01%). The IR spectrum showed absorption bands due to the end-methylene (v_{max} (in Nujol) 3077, 1638 and 888 cm $^{-1}$) and the carboxyl groups (3300~2800 and 1708 cm $^{-1}$). The NMR and MS spectra demonstrated the presence of the isopropenyl group ($\delta_{
m ppm}$ (in CDCl $_3$) 1.75 (bs, 3H), 4.70 (bs, 1H) and 4.85 (bs, 1H); m/e 41). The $^{13}C-NMR$ spectrum in CDCl $_3$ also indicated the presence of the carboxyl (δ_{ppm} 180.2), the end-methylene (147.4 and 113.4) and the ether-ring carbons (85.8 and 85.3). The peaks at m/e 141 (base), 123 and 43 in the MS spectrum of alnuseric acid was quite similar to those in it of alnincanone (II), a naturally occurring dammarane-type triterpene with a tetrahydrofuran ring. 6,7) Ryabinin et al. 6) has proved the occurrence of such m/e 141, 123 and 43 peaks from the tetrahydrofuran ring of alnincanone (II). also been established in the MS spectra of almuserol (III) 4) and various compounds 8,9) Therefore, the occurrence of these peaks in the MS with a tetrahydrofuran ring. spectrum of alnuseric acid (Scheme I) demonstrated that this acid possesses the side chain with a tetrahydrofuran ring similarly to II and III.

$$m/e \ 413$$
 $m/e \ 413$
 $m/e \ 414$
 $m/e \ 414$
 $m/e \ 414$
 $m/e \ 416$
 $m/e \ 417$
 $m/e \ 418$
 $m/e \ 419$
 $m/e \ 419$

Scheme I. Mass spectral fragmentation pattern of alnuseric acid (I).

The other interesting feature of alnuseric acid is that it contains no carbonyl or hydroxyl group at C-3, though such functional groups appear almost without exception in plant triterpenoids. In addition, the molecular formula indicates that alnuseric acid has four rings whereas alnincanone (II) and alnuserol (III) have five rings. These structural features of alnuseric acid may result from the cleavage of a 3-keto-triterpenoid. To the alnuseric acid, we here propose a biogenetically acceptable structure (I) 5) probably derived from alnincanone or its derivatives by

ring-opening at the 3,4-position. The proposed structure (I) possesses all the features necessary to explain the spectral data of alnuseric acid.

The structure (I) proposed for alnuseric acid was confirmed by its synthesis starting from alnincanone following the method established for ring-opening at the 3,4-position in the A ring of β -amyrenone. Alnincanone (II) was transformed to the corresponding ketoxime, which then was treated with p-toluenesulfonyl chloride in dry pyridine to yield the abnormal Beckmann rearrangement product (IV) with infrared bands (in Nujol) at 2240 (-C=N) and 3075, 1632 and 899 cm⁻¹ (>C=CH₂). The mild alkaline hydrolysis of the product (IV) gave (20s,24r)-20,24-epoxy-24-methyl-3,4-secodammar-4(28)-en-3-oic acid (mp 107~109°C; m/e 472 (M⁺), 141 (base), 123, 43, 41; $\nu_{\rm max}$ (in Nujol) 3300~2800, 3079, 1710, 1638, 890 cm⁻¹; $\delta_{\rm ppm}$ (in CDCl₃) 1.75, 4.70, 4.85). Identity of naturally occurring alnuseric acid with this authentic acid was established by comparison of mixed mp, TLC, IR, NMR, and MS.

Thus, the structure of alnuseric acid (I) has been elucidated to be (20s,24R)-20,24-epoxy-24-methyl-3,4-secodammar-4(28)-en-3-oic acid. To our knowledge, the acid I is the first reported naturally occurring C₃₁-3,4-secodammarane-type triterpene. Along with the acid I, alnincanone (II) has also been found in the male flowers cf Alnus serrulatoides. It is fascinating to note that II might be an intermediate for the biosynthesis of the new seco-triterpene I.

Acknowledgment. The authors thank to Dr. T. Hinomoto of JEOL Co. Ltd. for measuring the ¹³C-NMR spectra. One of the authors (T.S.) is also grateful to the Ministry of Education for financial support of this investigation through Grant-in-aid (047089) for Scientific Research in 1975.

References

- 1) J. Heslop-Harrison, "Pollen: Development and Physiology," London Butterworths Co. Ltd., London (1971).
- 2) T. Suga, T. Hirata, and N. Iwata, Chem. Lett., 971 (1974).
- 3) T. Hirata, K. Murai, R. Ideo, and T. Suga, The pre-print of "The 20th Symposium on the Chemistry of Natural Products," Sendai (1976) p.273.
- 4) T. Hirata, K. Murai, T. Suga, and A. Christensen, Chem. Lett., 95 (1977).
- 5) A. A. Newman, "Chemistry of Terpenes and Terpenoids," Academic Press Inc., New York, N.Y. (1972) p.214.

- 6) A. A. Ryabinin, L. H. Matyukhina, I. A. Saltikova, F. Patil, and G. Ourisson, Bull. Soc. Chim. Fr., 1089 (1968).
- 7) R. Labriola and G. Ourisson, Tetrahedron, 29, 2105 (1973).
- 8) Q. N. Porter and J. Baldas, "Mass Spectrometry of Heterocyclic Compounds," Wiley-Interscience Inc., New York, N.Y. (1971) p.39.
- 9) H. Budzikiewicz, C. Djerassi, and D. H. Williams, "Mass Spectrometry of Organic Compounds," Holden-Day Inc., San Francisco, Calif. (1967) p.227.
- 10) G. H. Whitham, J. Chem. Soc., 2016 (1960).

(Received January 24, 1977)